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DEPARTMENT OF ECOLOGY

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M E M O R A N D U M July 22, 1985

To:

Tom Eaton

Through:

Bill Yake D

From:

Art Johnson $\mathcal{A}/$

Subject:

Receiving Environment Survey in Budd Inlet at McFarland/Cascade,

February 13, 1985

ABSTRACT

Samples of creosote, ground water, marine discharges, and intertidal sediment at the McFarland/Cascade wood-treating facility in Olympia, Washington, on Budd Inlet were analyzed to determine the nature and extent of contamination by creosote and pentachlorophenol. Aromatic hydrocarbons indicative of creosote were detected in all media. Pentachlorophenol was detected in water only. The intertidal sediments near McFarland/Cascade are potentially toxic to marine life. The concentrations of polynuclear aromatic hydrocarbon observed within the limited area of intertidal sediment surveyed are among the highest so far reported for Puget Sound.

INTRODUCTION

In response to your request, Dale Norton and I conducted a survey on February 13, 1985, in the intertidal zone around McFarland/Cascade's wood-treating facility in Olympia on Budd Inlet. The objectives of the survey were to determine if there were ongoing discharges of creosote, pentachlorophenol, or metals from this site and if there was evidence of sediment contamination. The survey was intended to be preliminary in nature and result in recommendations for additional work, if warranted. The results of our survey follow.

SAMPLING

Figure 1 shows the locations where samples were collected. Three discharges were sampled; McFarland/Cascade's stormwater discharge to East Bay (NPDES No. WA-000101-5), a bank seep near the permitted discharge, and an off-site storm drain discharging to West Bay. The West Bay discharge was sampled because it is in communication with contaminated ground water beneath McFarland/Cascade. [This communication was apparently established when a drain for the East Bay Marina project was dug along the McFarland/Cascade south property line in 1983 (R. Pierce, personal communication). A December 1983 water sample from the West Bay discharge (lab no. 40566) collected by the WDOE Southwest Regional Office and analyzed at the EPA/WDOE Manchester laboratory had a polynuclear aromatic hydrocarbon (PNA) concentration of 193,000 ug/L.]

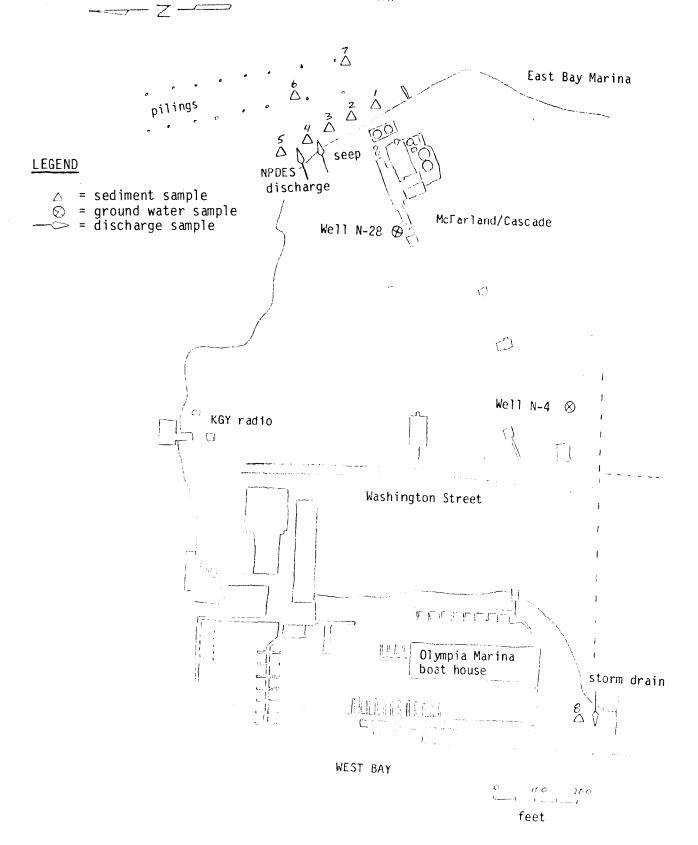


Figure 1. Locations of samples collected by WDOE on February 13, 1985, at McFarland/Cascade's wood-treating facility on Budd Inlet.

Seven sediment samples were collected in the East Bay intertidal zone. Five of these were in a transect about 20 yards off the riprap bank that forms the northeast border of McFarland/Cascade. Two more sediment samples were collected outside this transect, about 50 yards from the bank. One sediment sample was also taken in West Bay next to the outfall of the above-mentioned storm drain.

To better relate results from analysis of these samples to the substantial onsite contamination already documented in the preliminary ground water evaluation (Applied Geotechnology, 1984), samples of ground water from monitoring wells designated N-4 and N-28, and a sample of creosote from Tank No. 2 were also collected. These wells were selected because of the high contaminant concentrations found by Applied Geotechnology and their proximity to the discharges and sediments being sampled. This shallow aquifer is apparently completely isolated from local water supply wells (Applied Geotechnology, 1984).

Water was collected as grab samples in one-gallon glass jars with teflon-lined lids (priority pollutant acid/base-neutral compounds), 40 mL glass vials sealed with teflon septums (priority pollutant purgeable organics), one-pint polyethylene cubitainers (copper, chromium, arsenic), one-half-liter polyethylene bottles (pH, specific conductivity, total suspended solids), and 250 mL polyethylene bottles with 2 mL H₂SO₄ as preservative (total organic carbon). Flows were instantaneous measurements taken with a Marsh-McBirney magnetic flow meter, or, alternatively, with a bucket and stopwatch. (Because of analytical problems experienced at the EPA/WDOE Manchester laboratory, the metals data on discharges are from a second set of water samples collected April 24, 1985.)

The ground water wells were purged by pumping with a peristaltic pump and teflon tubing until specific conductivity stabilized. This required removing about four well volumes from each well. New tubing was used for each well. Samples were taken with 1000 cc teflon bailers (Johnson UOP) on monofilament lines. The distance from the top of the well casing to the ground water surface was measured with an E-tape.

Sediment samples were collected from the top one foot of sediment with large stainless steel spoons. Sample containers were eight-ounce glass with teflonlined lids (acid/base-neutrals and purgeables) and small polyethylene cups (copper, chromium, arsenic, total organic carbon, and percent solids).

Containers for priority pollutant organics were cleaned with sequential rinses of detergent, HCl, HNO3, distilled water, nanograde acetone, and nanograde methylene chloride, then dried at 350°F for 24 hours. Well bailers and spoons used in sediment sampling were washed with detergent and rinsed with methylene chloride before use. Samples were kept on ice after collection. Chain-of-custody procedures were followed.

ANALYSIS

Analysis was done at the EPA/WDOE Manchester laboratory except for total organic carbon in sediment which was done by Laucks Testing Laboratories, Inc., Seattle.

Priority pollutant organics were analyzed by gas chromatography/mass spectrometry following EPA method 625 (acid/base-neutrals) and EPA method 624 (purgeables). Priority pollutant analysis included compounds on the EPA Hazardous Substances List. Additional organic compounds were tentatively identified by computer-match with the Manchester laboratory spectrum library. Metals were analyzed by atomic absorbtion spectrometry following EPA methods 220.1 (copper), 218.1 (chromium), and 206.2 (arsenic). Sediment samples were digested with HNO3 and H2O2 for metals analysis. Unfiltered water samples were analyzed for total metals. No analyses were done for priority pollutant pesticides, PCBs, or dioxins. Dickey Huntamer (WDOE) did the acid/base-neutrals analysis, Steve Pope (EPA) did the purgeables analysis, and Roy Araki (WDOE) did the metals analysis.

Analysis for conventional parameters in water and percent solids in sediment followed EPA (1979) Methods for Chemical Analysis of Water and Wastes. Total organic carbon in sediment was determined by loss on ignition.

Field blanks were analyzed for both the priority pollutants organics (lab. nos. 079014 and 079015) and metals (lab. no. 131454). Blank water was prepared at the EPA/WDOE Manchester laboratory by passing de-ionized water through activated carbon filters. All constituents analyzed were below detection limits in field blanks. Laboratory blanks for some of the purgeables analyses contained methylene chloride, acetone, and 2-butanone.

Spike recoveries for the organics analysis were within limits specified for the EPA Contract Laboratory Program except for the surrogate spikes 2-fluorophenol, D-5 phenol, and D-5 nitrobenzene. This problem is currently being investigated by the analyst.

RESULTS AND DISCUSSION

The results of conventional and priority pollutant analyses are in Table 1.

The analyses done on creosote showed it to contain high concentrations of a number of organic compounds classed as priority pollutants or hazardous substances. A major fraction of the creosote was low molecular weight PNAs †

[†]In this report, low molecular weight PNAs refer to the 2- and 3-ring compounds naphthalene, 2-methylnaphthalene, acenaphthylene, acenaphthene, fluorene, phenanthrene, and anthracene; high molecular weight PNAs refer to the 4-ring and above compounds fluoranthene, pyrene, benzo(a) anthracene, chrysene, benzoflouranthenes, benzo(a)pyrene, indeno(1,2,3-c,d) pyrene, and benzo(q,h,i)perylene.

Table 1. Results of analysis of product, ground water, site discharges, and intertidal sediment samples collected by WDOE at McFarland/Cascade wood-treating facility on Budd Inlet, February 13, 1985 (ppb unless otherwise noted).

Sample				NPDES	1 69	Storm	4 4 5 1	1000	7 7 7			East Ba	sy Off-	W. Bay
	Creacate	We11	Well	to t	t (to L	Sta.	Sta.	Sta.	Sta.	Sta.	Sta.	St a.	St a.
Laboratory Number	079024	2106/0	079009	0/9010		7. 8ay	10	710670	#3 079018	#4 079019	#5 079020	#6 07902T	#7 079022	#8 079023
Conventional Parameters flow (cfs) temperature (*C) pH (units) spec. Cond. (umhos/cm) t. susp. solids (mg/L) t. organic carbon (%) percent solids	# 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4	9.6 6.3 778 850 N/A	10.8 7.1 1,820 310 N/A	0.007 10.3 7.7 218 14 N/A	0.004 6.8 6.9 23,400 12 N/A	0.33 9.1 7.5 877 29 N/A	3.5 45.0	3.2 47.6	3.8 49.9	1.9 67.1	3.8 67.2	2.4 57.3	1.6 52.4	4.2 70.2
Aromatic Hydrocarbons benzene ethylbenzene toluene total xylenes naphthalene acenaphthalene acenaphthene dibenzofuran fluorene phenanthrene anthracene fluoranthene pyrene benzo(a)anthracene chrysene benzo(a)pyrene benzo(a)pyrene benzo(a)pyrene benzo(a)pyrene benzo(a)pyrene benzo(a)pyrene	13,000 61,000 58,000 120,000 82,000,000 3,000,000 3,000,000 24,000,000 24,000,000 80,000,000 80,000,000 5,700,000 7,700,000 4,000,000 1,100,000 610,000	263 380 210 210 700 21,000 25,400 67 11,100 11,100 11,100 11,000 226 255 255 250 210 210 210 210 210 210 210 210 210 21	150 90 90 90 600 11,500 117 117 117 117 117 117 117 117 117	288 4 288 111 22 888 888 888 888 888 888 888	50 50 50 50 60 60 60 60 60 60 60 60 60 60 60 60 60	400 1123 1123 1123 1123 1120 1120 1130 118 118 118 118 118 118 118 118 118 11	9U 118 9U 9U 730 730 5,300 5,300 20,000 22,000 22,000 8,000 8,000 8,000 1000 1000	5:0 2:0 400 2:0,000 115,000 370,000 370,000 130,000 150,000 150,000 95,000 120,000 40,000 40,000 4000	70 70 70 70 70 70 70 70 840 11,400 11,400 7,900 7,900 7,500 7,500 7,500 7,500 7,500 7,500 7,500 7,500 7,500 7,500 8,700 7,500 8,700	50 50 50 50 50 640 11,500 11,500 310 650 650 650 670 670 670 670 670 670 670	50 50 50 50 50 50 50 50 62 62 62 62 62 62 63 64 64 64 64 64 64 64 64 64 64 64 64 64	70 70 70 70 70 70 70 70 70 70 70 70 70 7	55 55 55 530 1100 1100 1100 11,300 11,300 11,800 11,800 11,000	50 50 50 50 50 1000 1000 1000 1000 1000
Phenols phenol 2-methylphenol 4-methylphenol 2,4-dimethylphenol pentachlorophenol	1,000,000 500,000 1,400,000 510,000J 25,000U	3,600 510 1,900 20 1,900	11 11 22 92	10 10 10 20 1,700	0.1U 0.1U 0.1U 0.1U 8.6	10 10 10 20 20 17	1000 1000 1000 1000	4000 4000 4000 4000 4000	400U 400U 400U 400U 400U	1000 1000 120 1000 1000	1000 1000 1000 1000	2000 2000 2000 2000 2000	1000 1000 1000 1000	1000 1000 1000 1000
Miscellaneous Organics bis(2-ethylhexyl) phthalate chlorobenzene 2-butanone carbon disulfide 2-methyl-2-pentanone methylene chloride acetone	25,000U 9,000U 54,000 9,000U 6,000J 12,000 39,000	85U 40U 110B 40U 80U 84B	230 50 50 50 50 50 30 30 700	30 50 50 50 50 100 30 50	0.6U 50 6JB	30 400 900 400 800 2208	760 90 24 91 1180 2708	15,000 8U 38 43 16U 16U 2008	1,100 70 27 70 70 140 16JB	1000 50 1100 50 50 908 35	120 50 50 50 50 100 50 33	640 7U 114U 7U 11U 11JB	420 5t 22u J 22 23 33 33 120u J	1,600 10 10 41UJ 50 10U 14J 87UJ
Metals copper chromium arsenic	N/A N/A N/A	154 123	84 8 N/A 5	21+ 1 27+ 6 31+ 1	14+ 1 6+ 1 15+ 6	14† 13† 6†	38.377 38.877 3.077	69,611 35,011 5,711	131.5†† 40.2†† 10.7††	37.4†† 20.5†† 11.0††	32.0†† 20.2†† 5.5††	62.611 35.211 7.611	71.5 11 35.2 11 7.311	56.61† 20.61† 9.01†

NOTE: Sediment data are on a dry-weight basis. #Metals data on discharges are from samples collected May 24, 1985. #TSediment netals units are ppm.

N/A = not analyzed J = estimated corcentration U = not detected at detection limit shown B = analyte founc in laboratory blank as well as sample

which comprised 28 percent of the sample. Lesser amounts of high molecular weight PNAs (8.5 percent), phenol/cresols (0.3 percent), and single-ring aromatic hydrocarbons (0.03 percent) were measured. Two ketones (2-butanone and 2-methylpentanone), methylene chloride, and acetone were also detected as minor constituents. The predominant compounds among the priority pollutants analyzed were the low molecular weight PNAs naphthalene and phenanthrene, each present at about 8 percent.

The water level in well N-28 was 51 1/2 inches below the top of the casing; the level in well N-4 was 93 inches. Based on data in the Applied Geotechnology report, the water elevation in well N-28 was 14.3 feet above mean lower low water. Casing elevation was not reported for well N-4.

There was evidence of substantial ground water contamination by creosote and pentachlorophenol in both wells. Well N-28, near the East Bay shoreline, had much higher contaminant levels than well N-4. During the process of sampling. a large amount of solids became suspended in well N-28 and incorporated into the samples. Although the N-28 samples were allowed to settle overnight and then decanted prior to analysis, the data for this well may overestimate concentrations of some constituents, especially PNAs which tend to be associated with particulates.

Naphthalene was the predominant PNA in ground water--well N-28 had 21,000 ug/L; well N-4 had 5,000 ug/L. Pentachlorophenol concentrations were 1,900 ug/L in well N-28 and 92 ug/L in well N-4. Applied Geotechnology also found naphthalene to be the predominant PNA in ground water. They reported 14,000 ug/L naphthalene in well N-28 and 11,600 ug/L naphthalene in well N-4 (analysis by Laucks Testing Laboratories, Inc., Seattle). Their pentachlorophenol analysis included other chlorinated phenols and is not, therefore, comparable with results from the present survey. The two data sets are compared below in Table 2. Applied Geotechnology's analyses were limited to PNAs, chlorophenols, and oil and grease.

Table 2. Comparison of WDDE and Applied Geotechnology, Inc. data on McFarland/Cascade monitoring wells N-4 and N-28.

Well Number	*****	N-4		N-28
Investigator	WDOE	AGI	WDOE	AGI
Sample Date	2/13/85	11/6/83	2/13/85	12/7/83
naphthalene (ug/L)	5,000	11,600	21,000	14,000
2-methyl naphthalene (ug/L)	1,500	N/A	5,400	N/A
acenaphthylene (ug/L)	13	2,060	67	160
acenaphthene (ug/L)	320	1,560	2,500	239
dibenzofuran (ug/L)	1J	N/A	1,100	N/A
fluorene (ug/L)	58	168	1,200	100
phenanthrene (ug/L)	69	152	3,000	106
anthracene (ug/L)	17	4.21	430	14
fluoranthene (ug/L)	17	142	1,100	1.40
pyrene (ug/L)	10	13	1,000	10
benzo(a)anthracene (ug/L)	4	3.53*	240	4.5*
chrysene (ug/L)	4J		250	
benzofluoranthenes (ug/L)	2U	6.00**	160	19U*
benzo(a)pyrene (ug/L)	2U	3.29	60	9U
indeno(1,2,3-cd)pyrene (ug/L)	20	3.60	20	290
benzo(g,h,i)perylene (ug/L)	20	5. 4 U	20	43 U
dibenzo(a,h)anthracene (ug/L)	N/A	9.80	N/A	78U
pentachlorophenol (ug/L)	92	4,970†	1,900	19,900†
oil & grease (mg/L)	N/A	5 u	N/A	10

^{*}benzo(a)anthracene + chrysene

^{**}benzo(b)fluoranthene

tincludes other chlorophenols

N/A = not analyzed
J = estimated concentration

U = not detected at detection limit shown

McFarland/Cascade's NPDES discharge to East Bay had a high pentachlorophenol concentration of 1,700 ug/L. Low concentrations (4 ug/L or less) of xylene and six PNAs were also detected. The adjacent bank seep had 8.6 ug/L pentachlorophenol. The Manchester laboratory achieved lower detection limits for PNAs in the seep sample and was able to quantify a number of compounds at between 0.1 ug/L and 0.4 ug/L. Judging from its high specific conductivity, the seep represented primarily tidal waters. The storm drain to West Bay, unlike East Bay discharges, had a substantial PNA content ranging from 3 ug/L benzoflouranthenes to 150 ug/L acenaphthene. It also had a small amount of pentachlorophenol, 17 ug/L.

2-butanone, methylene chloride, and acetone were reported as present in ground water and discharge samples. These compounds were used in cleaning the sample containers and sampling equipment and/or were detected in laboratory blanks so their presence may be an artifact of sampling or analysis. (The concentrations of 2-butanone, methylene chloride, and acetone measured in the creosote sample, described earlier, are too large to attribute solely to contamination.)

Low concentrations of copper, chromium, and arsenic were characteristic of all ground water and discharge samples.

The pentachlorophenol concentration measured in the McFarland/Cascade discharge would require substantial dilution--32 to 50 fold--to be reduced below the EPA criteria of 53 ug/L (acute toxicity) and 34 ug/L (chronic toxicity) for protection of marine life (EPA, 1980). The total priority pollutant PNA concentration in the West Bay storm drain, 512 ug/L, is also above EPA's 300 ug/L criterion (acute toxicity) for these compounds.

Table 3 below summarizes the aromatic hydrocarbon and pentachlorophenol loads to East Bay and West Bay calculated from the February 13 survey. Measurement of the total contaminant load to Budd Inlet from the McFarland/Cascade site was not attempted in this survey. Since only one of the several existing seeps to East Bay was sampled, the loads in Table 3 underestimate the total loading for this time period. On the other hand, because the seeps and the West Bay storm drain are tidally influenced, average daily loads may be lower than indicated by these instantaneous measurements taken at low tide.

Table 3. Contaminant* loads to Budd Inlet from McFarland/Cascade and vicinity measured by WDOE on February 13, 1985.

Discharge	Constituent	Lo (1bs	ad /day)
Storm Drain to West Bay	low molecular weight PNAs high molecular weight PNAs single-ring aromatics pentachlorophenol	0. 0. 0.	15 10
McFarland NPDES discharge to East Bay	low molecular weight PNAs high molecular weight PNAs single-ring aromatics pentachlorophenol	0.	0001 0003 0002 06
McFarland Seep to East Bay	low molecular weight PNAs high molecular weight PNAs single-ring aromatics pentachlorophenol	0. not de	00002 00002 tected 0002

^{*}Organics loads do not include tentatively identified compounds

The West Bay storm drain had several orders of magnitude higher aromatic hydrocarbons loads than the other two discharges. The McFarland/Cascade permitted discharge was the predominant pentachlorophenol load. The West Bay aromatic hydrocarbon loads were 0.56 pound/day low molecular weight PNAs, 0.15 pound/day high molecular weight PNAs, and 0.10 pound/day single-ring compounds. The pentachlorophenol load from the McFarland/Cascade permitted discharge was 0.06 pound/day. Based on results of the preliminary ground water evaluation, Applied Geotechnology estimated a load of 10 pounds/year of combined PNAs and chlorophenols in ground water going to Budd Inlet from McFarland/Cascade. However, an error was made in calculating this load (Applied Geotechnology, 1984). A better estimate, although still preliminary, is on the order of 450 pounds/year or about 1 pound/day (M. Adams, personal communication).

The results of sediment sample analysis showed very high PNA concentrations—up to 800,000 ug/Kg individual PNAs—in the East Bay transect near McFarland/Cascade, reduced PNA concentrations in the two East Bay samples farther off—shore, and only two individual PNAs detectable in the West Bay sample. The low PNA level in the West Bay sample in spite of a nearby source is puzzling and suggests additional samples should be collected here.

Phenol, cresols, and pentachlorophenol were not detected in sediment except for 120 ug/Kg of 4-methylphenol in one of the East Bay transect samples. This may be a function of the relatively high solubility of these compounds and/or the detection limits employed (100 to 400 ug/Kg). Puget Sound background levels for phenols and cresols are below 100 ug/Kg (Tetra Tech, in prep.).

Six other organic priority pollutant compounds were detected in sediment. Bis(2-ethylhexyl)phthalate, a widely used plasticizer, was quantified at between 120 and 15,000 ug/Kg in six of the eight samples collected. Small amounts (<100 ug/Kg) of chlorobenzene and carbon disulfide were detected in several samples. The significance of 2-butanone, methylene chloride, and acetone in sediment is questionable for reasons already explained.

The copper, chromium, and arsenic levels measured in the East and West Bay sediments, although elevated, are not unusual for a Puget Sound urban embayment.

Figure 2 depicts the horizontal distribution of low and high molecular weight PNAs in East Bay intertidal sediments near McFarland/Cascade. Concentrations peak at station #2 near the wood-treatment facility and decrease moving both to the northwest in the nearshore transect and in the offshore direction. Correcting these data for total organic carbon does not alter the direction of these gradients, indicating the observed distribution is not a simple function of carbon content, and strongly suggesting the source is McFarland/Cascade. Field observations recorded during sampling noted an oily sheen and droplets of creosote-like material at stations #1, #2, and #3 where PNA concentrations were found to be greatest.

The East Bay sediment data were compared to a summary of information on Puget Sound "toxic hot spots" (Yake, 1985). The station #2 sample at McFarland/ Cascade had an especially large concentration of high molecular weight PNAs and, at 1,300 mg/Kg, is substantially more contaminated than the highest

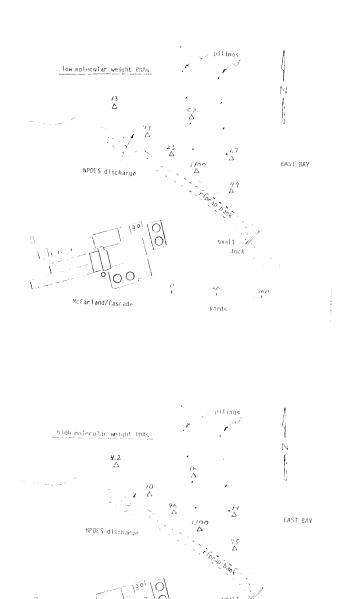


Figure 2. Horizontal distribution of PNAs in East Bay intertidal sediments adjacent to McFarland/Cascade's wood-treating facility; based on samples collected by WDOE, February 13, 1985 (mg/Kg, dry weight).

mcFarland/Cascade

samples from Hylebos Waterway, Eagle Harbor, or Harbor Island -- the top three PNA hot spots known in Puget Sound (308 to 350 mg/Kg maximum high molecular weight PNA concentrations observed). In the limited area covered by the present survey--approximately 1,000 yd 2 --it appears that PNA concentrations drop off by one to two orders of magnitude within 50 to 100 yards of station #2. Stations #1 and #3 remain sufficiently high to be classed as PNA hot spots.

The levels of PNAs and dibenzofuran measured in these East Bay sediment samples may be sufficient to harm marine life. Table 4 compares the East Bay and West Bay data with draft "apparent effects thresholds" determined by Tetra Tech, Inc. during the recent Superfund investigation of the nearshore/tideflats area of Commencement Bay. These thresholds were derived by comparisons of biological response to sediment chemistry (as measured by amphipod and oyster larvae bioassays and abundance of infauna) at stations where these data were obtained concurrently. They represent a concentration which appears likely to have deleterious effects on marine organisms. As shown in Table 4, only the West Bay sample is below apparent effects thresholds in all categories.

Table 4. Summary of PNA data on intertidal sediments near McFarland/Cascade showing potential for adverse effects on marine life $(\frac{ug}{v_0}/Kg, dry)$.

		_	-	East Bay	Nearshore	Transect		East Bay	Offshore	West Bay
"Apparent	Effects	Thresholds"a	Sta. #1	Sta. #2	Sta. #3	Sta. #4	Sta. #5	Sta. #6	Sta. #7	Sta. #8
1mw PNAs	5.2	(diversity & toxicity)	<u>/49/</u>	/1700/	<u>/23/</u>	<u>/9.1</u> /	/13/	<u>/5.2/</u>	1.7	0.10U
hmw PNAs	12	(reduced infaunal diversity)	<u>/75</u> 7	/13007	<u>/967</u>	10	4.2	/16/	<u>/14/</u>	0.88
hmw PNAs	17	(toxicity in amphipod/oyster larvae bioassays)	<u>/75</u> /	/1300/	<u>/96</u> 7	10	4.2	16	16	0.88
dibenzofur		(diversity & toxicity)	<u>/5.3/</u>	/1307	/1.47	0.31	0.25	0.72	0.08J	0.100

^aDraft findings, Commencement Bay Nearshore/Tideflats Remedial Investigation (Tetra Tech, Inc., in prep.).

An additional 55 organic compounds, shown in Table 5, were tentatively identified in samples collected during the present survey. These compounds were primarily substituted aromatic hydrocarbons, especially benzenes and naphthalenes. The concentrations estimated for a number of these compounds equaled or exceeded those for the priority pollutants previously discussed.

^{/7 =} above threshold

U = not detected at detection limit shown

J = estimated concentration

Table 5. Compounds tentatively in product, ground water, site discharges, and intertidal sediment samples collected by WODE at McFarland/Cascade wood-treating facility on Budd Inlet, February 13, 1985 (ppb).

				NPDES Disch.	Seep	Storm Drain	East B	ay Nearsh	ore Sed				ediment	W. Bay Sedimen
Sample	Creosate	Well N-28	Well N-4	to E. Bay	to F. Bay	to W. Bay	5ta.	Sta. 12	Sta. #3	Sta. #4	Sta. #5	Sta. #6	Sta. Ø7	Sta. #8
aboratory Number	079024	079012	079009	079010	079011	079013	079016	079017	079018	079019	079020	079021	079022	079023
,2,4-trimethylbenzene	680,000	1,100J	1,100J			150J		280J						**
-ethyl-2-methylbenzene	1,000,000	1,600J				73J		1,000J					~-	
opylbenzene		330J	463			44		300J		900				
,2,3,4-tetramethylbenzene		450	170			44		• • •		300				
(-methyl-2-cyclopen-1-yl)	***	~ **	1,000											
benzene	** **	5.500		***	1.2	530J				~~				
ethyl-4-methylbenzene methyl-2-propylbenzene		130					~ ~		~ ~	***				
,3-diethyl-5-methylbenzene		320			an. 100									
-but yn lbenzene	** ***					78	~ ~					~		
2,3-trimethylbenzene	** ***	~			-07 148	200J				00.3				
-ethyl-1,2-dimethylbenzene		~ ~					T- 14			80J			~ ~	
-propenalbenzene	· -				~ •	1,800J			* ~				~ =	
-ethylnaphthalene		2,200	210					120,000						
.5-dimethylnaphthalene	24,000,000		310			240	4,500	300,000		1,900			No. 400	
.2-dimethylnaphthalene*		5,500	79			44		••		2,900				
.4.6-trimethylnaphthalene		2,400				38			no. Mr					
,2,3,4-tetrahydronaphthalene		~-	***			23	12 000							
,3-dimothylnaphthalene			***	***			12,000 130	130,000		1,100	530			
,4-dimethylnaphthalene							100	780,000		2,800	2,100			
-methylnaphthalene								700,000			540	~ -		
-ethylnaphthalene											1,100			
,7-dimethylnaphthalene		500J				240J							₩~	
-methylnaphthalene .4-dihydro-1,4-methano	2.2							air ria						
naphthalene		63 0J						***						
-isocvanonaphthalene			76	~-		ao 10		•				~ ~		
-methyl-1-naphthylisocyanide		39			~~				~ ~					
H-cyclopenta[DEF]	20 202 200		16	••					44 100					
phenanthrene	38,000,000		15											
-methylanthracene*			10			26	10.000							
lH-benzo(a)fluorene			100 100	~-			15,000							
enzo[B]thiophene				**	31	290	~ *			1,800			~ ~	
i i benzoth i ophene	18,000,000					23		270,000						
,3,4-trimethylthiophene		440	27		2.1								~ -	
-methyl-benzo[B]thiophene*		440	240		1.6									
-methyl-benzo[8]thlophene		680			1.0									
nzulene			160			560J								
2,3-dihydroindene	AM 79													**
1,3-dihydro-4,7-dimethyl-			59											
lH-indene 2.3-dihydro-4-methyl-		***	F10			170					***			
1H-indene		980	510	***		170								
-ethylidene-1H-indene	110,000,000				~ ~									
2,3-dihydro-l-methyl-			F2.44			91J	***	~ =						
IH-indene						70J								
-methyl-lH-indene						703 48	5,800							
l-methyldibenzofuran			2,700			70	5,000							
7-methylbenzofuran	***		2,700			120J								
penzofuran														
[1',1'-biphenyl]-carboxy	9,300,000)	29			48								
aldehyde [,1'-biphenyl			470		** **	***			-	540	910	• •		
iphenylene			50		-									
isoquinolene						41	500							
arbazole	4,000,000)					680)						
2,3,4,6-tetrachlorophenol		. 180		140	2.9									
2-chloro-1,3,5-	190,000	3.1			ad 40					• ~				
cycloheptatriene	-					**								
3-phenyl-2-propenal	***	- 570	800			40			- *		•••			
alpha-methylbenzeni acetaldehyde		- 370					~-							
oxime(E)-4-methylbenz-			60		** **								***	
aldehyde														**
1-(methylphenyl)-ethanone		- 670				an				10.77				

MOTE: Sediment data are on a dry-weight basis.
-- = not detected
J = estimated value.
* = or isomer

SUMMARY

The major findings of this survey are as follows:

- Analysis of ground water in wells N-28 and N-4 confirmed earlier findings by Applied Geotechnology, Inc. of high PNA and pentachlorophenol concentrations.
- 2. PNAs and pentachlorophenol were detected in the McFarland/Cascade NPDES discharge, an adjacent seep, and a storm drain to West Bay. Concentrations of individual PNAs were in the range of 1 to 4 ug/L in the permitted discharge, 0.1 to 0.4 ug/L in the seep, and 3 to 150 ug/L in the storm drain; pentachlorophenol concentrations were 1,700 ug/L, 8.6 ug/L, and 17 ug/L, respectively.
- 3. The West Bay storm drain constituted the largest PNA load of the three discharges sampled. The low molecular weight PNA load was 0.56 pound/day, and the high molecular weight PNA load was 0.15 pound/day. Corresponding loads in the McFarland/Cascade permitted discharge and adjacent seep were several orders of magnitude lower.
- 4. The pentachlorophenol loads measured were 0.06 pound/day in the permitted discharge, 0.03 pound/day in the West Bay storm drain, and 0.0002 pound/day in the seep.
- 5. The concentrations of PNAs present in the West Bay storm drain and pentachlorophenol in McFarland/Cascade's permitted discharge would require dilution in the receiving waters to be reduced below levels considered toxic to marine life.
- 6. Evidence of creosote in the form of oily sheens and droplets of a creosotelike substance were observed in the East Bay intertidal zone near McFarland/ Cascade.
- 7. High concentrations of PNAs were found in East Bay intertidal sediments adjacent to McFarland/Cascade. Concentrations of high molecular weight PNAs ranged from 4.2 to 1,300 mg/Kg, with the highest levels occurring in those samples collected nearest the wood-treating facility. The higher concentrations measured are sufficient to rank this site among the Puget Sound PNA "hot spots." However, results of the present survey suggest the horizontal extent of contamination may be limited to a relatively small area.
- 8. The levels of PNAs found in East Bay intertidal sediments are potentially toxic to marine life.
- 9. Phenol and pentachlorophenol were not detected in sediment. One cresol (4-methyl phenol) was detected in a single sediment sample.

RECOMMENDATIONS

Short-Term:

- 1. Collect several additional sediment samples in West Bay to verify absence of significant creosote contamination.
- 2. Conduct an expanded sediment survey in East Bay to determine the horizontal and vertical extent of creosote contamination.
- 3. Analyze for PNAs and pentachlorophenol in samples from McFarland/Cascade seeps not included in the present survey.

Long-Term:

- 1. Eliminate on-going sources of contamination to Budd Inlet, probably through controlling contaminated ground water migrations from McFarland/Cascade.
- 2. Resolve sediment contamination, probably by requiring removal of all sediments exceeding the apparent effects threshold for any compound.

AJ:cp

Attachments

REFERENCES

- Applied Geotechnology, Inc., 1984. Preliminary Ground Water Evaluation, Cascade Pole Facility, Olympia, Washington.
- EPA, 1979. Methods for Chemical Analysis of Water and Wastes. EPA 600/21-79-020.
- EPA, 1980. Water Quality Criteria Documents; Availability. CFR. Vol. 45, No. 231.
- Tetra Tech, Inc., 1985. Commencement Bay Nearshore/Tideflats Remedial Investigation (in prep.).
- Yake, B., 1985. Compilation of Puget Sound "Toxic Hot Spot" Information on Puget Sound. WDOE Memorandum to D. Cunningham.